WHAT IS CLAIMED IS:

- 1. A process for the production of a 2-oxa-3-one androstane derivative, the process comprising reacting a 3-one androstane derivative with ozone to form a 2-oxa-3-one androstane derivative.
- 2. The process of claim 1, wherein the reaction is conducted in the presence of an organic or inorganic peroxide.
 - 3. The process of claim 2, wherein the peroxide is hydrogen peroxide.
- 4. The process of claim 1, wherein the reaction is carried out in a temperature range from about 1° C to about 50° C.
- 5. The process of claim 1, wherein the 3-one androstane derivative is reacted with ozone for about 3 hours to about 5 hours.
- 6. The process of claim 1, wherein the ozone that is reacted with the 3-one androstane derivative is present as a mixture of oxygen and ozone.
- 7. A process for the production of oxandrolone, the process comprising reacting mestanolone with ozone to form oxandrolone.
- 8. The process of claim 7, wherein the reaction is conducted in the presence of an organic or inorganic peroxide.
 - 9. The process of claim 8, wherein the peroxide is hydrogen peroxide.
- 10. The process of claim 7, wherein the reaction is carried out in a temperature range from about 1° C to about 50° C.
- 11. The process of claim 7, wherein the mestanolone is reacted with ozone for about 3 hours to about 5 hours.

- 12. The process of claim 7, wherein the ozone that is reacted with the mestanolone is present as a mixture of oxygen and ozone.
- 13. A process for the production of oxandrolone, the process comprising reacting mestanolone with ozone in the presence of hydrogen peroxide in a temperature range from about 1° C to about 50° C for about 3 hours to about 5 hours, the ozone being present in a mixture of oxygen and ozone.
- 14. The process of claim 13, wherein the mestanolone is present in an aqueous solution.
- 15. A process for the production of a 2-oxa-3-one androstane derivative, the process comprising reacting a 3-one androstane derivative of formula (IV)

$$\begin{array}{c}
CH_3 \\
CH_3
\end{array}$$

$$\begin{array}{c}
R^1 \\
R^3
\end{array}$$

$$\begin{array}{c}
R^3 \\
R^5
\end{array}$$

$$\begin{array}{c}
R^5
\end{array}$$

$$\begin{array}{c}
R^5
\end{array}$$

$$\begin{array}{c}
R^5
\end{array}$$

wherein R^1 , R^2 , R^3 , R^4 , R^5 , and R^6 are independently selected from the following group: hydrogen, C_{1-10} alkyl, C_{1-10} ketone, phosphate, C_{1-10} alkyl carboxylate, amino, hydroxy, thiol, C_{1-10} thioalkyl, C_{1-10} alkoxy, substituted C_{1-10} alkyl, and halogen;

with ozone to form a 2-oxa-3-one androstane derivative of formula (V)

wherein R¹, R², R³, R⁴, R⁵, and R⁶ are as above in formula (IV).

- 16. The process of claim 15, wherein the reaction is conducted in the presence of an organic or inorganic peroxide.
 - 17. The process of claim 16, wherein the peroxide is hydrogen peroxide.
- 18. The process of claim 15, wherein the reaction is carried out in a temperature range from about 1° C to about 50° C.
- 19. The process of claim 15, wherein the 3-one androstane derivative of formula (IV) is reacted with ozone for about 3 hours to about 5 hours.
- 20. The process of claim 15, wherein the ozone that is reacted with the 3-one androstane derivative of formula (IV) is present as a mixture of oxygen and ozone.
- 21. The process of claim 20, wherein the reaction is conducted in the presence of hydrogen peroxide.